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## CESIUM FLUORIDE PROMOTED FORMATION OF N-CHLORO AND N-BROMO AMINES AND IMINES

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Cesium fluoride is a very effective reagent for the fluoride promoted oxidation of nitriles and imines by  ${\rm Cl}_2$ ,  ${\rm Br}_2$  and  ${\rm BrCl}$ . Previous work by us had demonstrated that CsF fluoride was effective in promoting the oxidation of  ${\rm R}_{\rm f}{\rm CF=NF}$  by  ${\rm Cl}_2$  and  ${\rm Br}_2$  to  ${\rm R}_{\rm f}{\rm CF}_2{\rm NFCl}$  and  ${\rm R}_{\rm f}{\rm CF}_2{\rm NFBr}$ , the oxidation of  ${\rm R}_{\rm f}{\rm CN}$  to  ${\rm R}_{\rm f}{\rm CF}_2{\rm NCl}_2$  by  ${\rm Cl}_2$  and the apparent oxidation of  ${\rm R}_{\rm f}{\rm CN}$  to  ${\rm R}_{\rm f}{\rm CF}$  = NBr by  ${\rm Br}_2$  (S.C. Chang and D. D. DesMarteau, Inorg. Chem. 1983, 48, 771).

We have now made a detailed investigation of the CsF promoted oxidation of  $R_f$ CN, NCCF=NF,  $CF_2$ =NCl,  $(CN)_2$  and  $CF_3$ N=CFR $_X$  by  $Cl_2$ ,  $Br_2$  and BrCl. Significant results to be discussed will include the general high yield syntheses of  $R_X$ CF=NBr,  $CF_3$ N(Cl)CF $_2$ R $_X$ , a comparison of the reactivity of -C $_2$ N-vs--CF=NF in NCCF=NF, the stepwise halogenation of  $(CN)_2$  by  $Cl_2$  and  $Br_2$ , the novel compounds  $CF_3$ NBrCl and  $CF_3$ NBr $_2$  from  $CF_2$ =NCl and the unique CsF promoted conversion of  $R_f$ NCl $_2$  by  $Br_2$  to  $R_f$ NBrCl and  $R_f$ NBr $_2$ . Furthermore, the utility of BrCl-vs-Cl $_2$  in the formation of certain N-chloro compounds will be shown.